OIV-MA-BS-18 Analysis of alpha-diacarbonyl compounds by HPLC after derivation by 1,2diaminobenzene in spirit drinks of viti-vinicultural origin Method OIV-MA-BS-18 : R2010

Type IV method

# Analysis of $\Box$ -dicarbonyl compounds in spiritous beverages of vitivinicultural origin by HPLC after derivation by 1,2 diaminobenzene

(OIV/OENO 382C/2010)

# 1. Introduction

The principal  $\square$ -dicarbonyl compounds found in wine-based spirits (Figure 1) are: glyoxal, methylglyoxal, diacetyl and pentane-2,3-dione.

Glyoxal	OCH-CHO (ethanedial)	
Methylglyoxal	CH <sub>3</sub> -CO-CHO (2-oxopropanal)	
Diacetyl	CH <sub>3</sub> -CO-CO-CH <sub>3</sub> (butane-2,3-dione)	
Pentane-2,3-dione	CH <sub>3</sub> -CH <sub>2</sub> -CO-CO-CH <sub>3</sub>	
Hexane-2,3-dione	CH <sub>3</sub> -CH <sub>2</sub> -CH <sub>2</sub> -CO-CO-CH <sub>3</sub>	
Figure 1 The principal a-dicarbonyl compounds of wine-based spirits (beyane-2.3-		

Figure 1. The principal  $\alpha$ -dicarbonyl compounds of wine-based spirits (hexane-2,3-dione is not naturally present in wine but it is used as internal standard).

Dicarbonyl compounds are important because of their sensory impact,

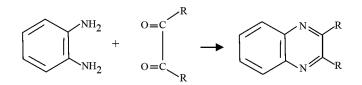
# 2. Applicability

This method applies to spirituous beverages of vitivinicultural origin for dicarbonyl compounds with a content ranging between 0.05 mg/L and 20 mg/L;

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diaminobenzene in spirit drinks of viti-vinicultural origin **Principle** 

The method is based on the formation of quinoxaline derivatives from  $\square$  - dicarbonyl compounds with 1,2-diaminobenzene (figure 2).



1,2-Diaminobenzene Dicarbonyl Quinoxaline

Figure 2 Formation of derivatives.

3.

The reaction takes place in the spirituous beverage diluted four-fold, pH 8 and after a reaction time of 3 hrs at 60° C. The analysis of the derivatives is then carried out either directly by chromatography in the high-performance liquid phase (HPLC) and detection by UV absorptiometry at 313 Nm,.

# 4. **Reagents and products**

#### 4.1. Dicarbonyl compounds

4.1.1.	Glyoxal (CAS N° 107-22-3) in a 40% solution
4.1.2.	Methylglyoxal (CAS N $^{ m o}$ 78-98-8) in a 40% solution
4.1.3.	Diacetyl (CAS N° 431-03-8) > 99 % pure
4.1.4.	Pentane-2,3-dione (CAS N $^{\circ}$ 600-14-6) > 97 % pure
4.1.5.	Hexane-2,3-dione (CAS N° 3848-24-6) > 90 % pure

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#### diaminobenzene in spirit drinks of viti-vinicultural origin 4.2. 1,2-Diaminobenzene (CAS N° 95-54-5) in the form of powder, > 97 % pure

- 4.3. Water for HPLC (according to standard EN ISO 3696)
- 4.4. Ethanol (CAS N° 64-17-5) pure for HPLC
- 4.5. Sodium Hydroxide (CAS N $^{\circ}$  1310-73-2) in 0.1M solution
- 4.6. Acetic acid (CAS N° 64-19-7) pure crystallisable

# 4.7. Solvent A for the analysis by HPLC

In 1 water L for HPLC (4.3), add 0.5 ml of acetic acid (4.8), mix, degas (by ultrasound, for example)

# 4.8. Solvent B for HPLC

Pure HPLC methanol (CAS N° 67-56-1)

# 4.9. 50% vol. hydroalcoholic solution.

Mix 50 ml of pure ethanol for HPLC (4.4) with 50 ml of water (4.3)

# 4.10. Solution of internal standard hexane-2,3-dione at 2.0 g/L $\,$

Place 40 mg of hexane-2,3-dione (4.2) in a 30 ml flask, dilute in 20 ml of 50% vol. hydroalcoholic solution. (4.11), stir until complete dissolution.

# 5. Apparatus

# 5.1. High-performance liquid phase chromatography with

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diaminobenzene in spirit drinks of viti-vinicultural origin detection by UV absorption (313 nm);

- 5.1.1. Analytical column filled with silica grafted by octadecyl radicals of 5 µm with dimensions of 250 mm x 4.6 mm, for example.
- 5.1.2. Data acquisition system.
- 5.2. pH measuring apparatus
- 5.3. Magnetic stirrer
- 5.4. Mg analytical balance
- 5.5. Solvent degasification system for HPLC (an ultrasound apparatus, for example)
- 5.6. Oven which can be set to 60°C
- 5.7. Standard laboratory glassware including pipettes, 30-ml (5.7) screw-cap flasks, and microsyringes.

# 6. **Preparation of the sample**

Dilute the spirituous beverage four-fold in water (4.3)

# 7. **Procedure**

Place 10 ml of spirituous beverage diluted four-fold (6) in a 30 ml flask Bring to pH 8 while stirring, with sodium hydroxide 0.1 M (4.5) Add 5 mg of 2,3-diaminobenzene (4.2) Add 10 µl of hexane-2,3-dione (internal standard) at 2.0 g/1 (4.10)

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diaminobenzene in spirit drinks of viti-vinicultural origin Close the flask using a screw-cap fitted with a Teflon-faced seal

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Stir until the reagent has completely disappeared (5.3)
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Place in the oven at 60°C for 3 hrs (5.6)

Cool.

# 7.1. Analysis

Injection. After cooling, the reactional medium containing the quinoxalines is directly injected into the HPLC system at an amount of 20  $\mu$ l.

• *Elution programme.* For separation, an example of an elution schedule is displayed in Table 1

Table 1. Example of HPLC analysis elution schedule			
Time in minutes	Solvent A	Solvent B	
0	80	20	
8	50	50	
26	25	75	
30	0	100	
32	0	100	

The flow rate being 0.6 ml/min

- Separation. The chromatogram obtained by HPLC is shown in Figure 3
- *Detection*. The maximum absorption was studied for all the dicarbonyl compound derivatives and set at 313 Nm as being optimal.
- *Identification of the derivatives.* The identification of the derivatives was carried out by comparing the retention times with standard reference solutions. The chromatographic conditions enable a good separation of the peaks in all the wines.

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#### 7.1.1. Characteristics of the method

Some internal validation elements were determined but these are not a formal validation according the protocol for the planning, the implementing and the interpretation of the performance studies pertaining to analysis methods (OIV 6/2010)

• *Linearity*. The linearity of the method was tested using standard solutions (the hydroalcoholic solution at 12% vol. was used as a matrix) (Table 2). The quantitative analysis of the additions of dicarbonyl compounds showed that the method is linear for the four compounds with recovery rate varying between 92 and 117%.

Table 2. Study of the linearity and recovery tests with standard solutions (12% v/v water-ethanol) correlation coefficients			
Glyoxal	Methylglyoxal	Diacetyl	Pentane-2,3-dione
value <sup>a</sup> surface <sup>b</sup>	value <sup>a</sup> surface <sup>b</sup>	value <sup>a</sup> surface <sup>b</sup>	value <sup>a</sup> surface <sup>b</sup>
R=0,992	R=0,997	R=0,999	R=0,999

a: mg/l, b: arbitrary units, c: response factor in relation to the internal standard.

• *The quantification limit* of the dicarbonyl compounds is very low, the best results being obtained with diacetyl, the detection limit of which is 10 times weaker than that of the other compounds (table 3).

*Tableau 3. Performance of the HPLC method for the quantification of dicarbonyl compounds* 

Limits	detection <sup>a</sup>	determination <sup>a</sup>	quantification <sup>a</sup>
Glyoxal	0,015	0,020	0,028

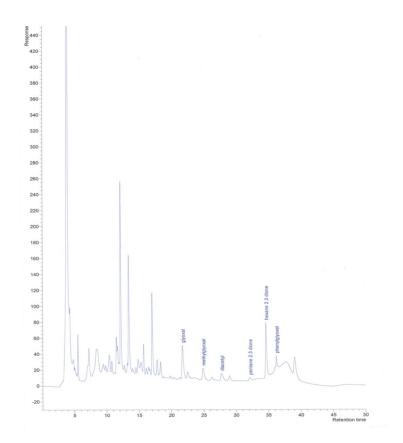
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Methylglyoxal	0,015	_0,020	0,027
Diacetyl	0,002	0,002	0,003
Pentane-2,3-dione	0,003	0,004	0,006

diaminobenzene in spirit drinks of viti-vinicultural origin

a: results in mg/L, hydroalcoholic solution (10% vol.).

Figure 3. High-performance liquid phase chromatogram of dicarbonyl compounds derivatized by 1,2-diaminobenzene, detected by UV at 313 nm. Spherisorb ODS Column 250 mm x 4.6 mm x 5  $\mu$ m.



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